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(54) Title: (AZAARYLMETHOXY)INDOLES AS INHIBITORS OF LEUKOTRIENE BIOSYNTHESIS

17 December 1992 (17.12.92)

Het
$$-X^4$$
 R^5
 R^5
 R^6
 $R^{11}R^{11})_{n^-}Y_{m^-}(CR^{11}R^{11})_{p^-}Q$
 R^6

(57) Abstract

4R2 (CA).

Compounds having formula (I) are inhibitors of leukotriene biosynthesis. These compounds are useful as anti-asthmatic, anti-allergic, anti-inflammatory, and cytoprotective agents. They are also useful in treating diarrhea, hypertension, angina, platelet aggregation, cerebral spasm, premature labor, spontaneous abortion, dysmenorrhea, and migraine.

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- 1 -

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TITLE OF THE INVENTION

(AZAARYLMETHOXY)INDOLES AS INHIBITORS OF LEUKOTRIENE
BIOSYNTHESIS

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BACKGROUND OF THE INVENTION

European Patent Applications 166,591 and 275,667 disclose a series of indole-based compounds with activity as prostaglandin antagonists and inhibitors of leukotriene biosynthesis respectively. In EP 181,568 and EP 200,101 are disclosed a series of compounds, containing two aromatic nuclei, which are described as possessing activity as lipoxygenase inhibitors. In EP 279,263 is disclosed a series of indoles, benzofurans and benzothiophenes which are described as possessing activity as lipoxygenase—

- 2 -

inhibitors. U.S. Patent 4,629,733 describes novel indolinones which are antithrombotic and inhibit both phosphodiesterase and tumor metastasis. The chemical preparation of quinolylindoles is referred to by Sheinkman, et al., Chem. Ab., Vol. 67, 54017 (1967), without mentioning any utility for such compounds. A number of N-acyl derivatives of indole-3-acetic acid are described as potential anti-inflammatory agents by Biniecki, et al., Chem. Ab., Vol. 98, 197936 (1983), by Pakula, et al., Chem. Ab., Vol. 105, 190835 (1986), and in British Pat. Spec. 1,228,848.

EP 419,049 (March 27, 1991) teaches (quinolin-2-ylmethoxy)indoles as inhibitors of leukotriene biosynthesis.

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SUMMARY OF THE INVENTION

The present invention relates to compounds having activity as leukotriene biosynthesis inhibitors, to methods for their preparation, and to methods and pharmaceutical formulations for using these compounds in mammals (especially humans).

Because of their activity as leukotriene biosynthesis inhibitors, the compounds of the present invention are useful as anti-asthmatic, anti-allergic, and anti-inflammatory agents and are useful in treating allergic rhinitis and chronic bronchitis and for amelioration of skin diseases like psoriasis and atopic eczema. These compounds are also useful to inhibit the pathologic actions of leukotrienes on the cardiovascular and vascular systems for example,

- 3 -

actions such as result in angina or endotoxin shock. The compounds of the present invention are useful in the treatment of inflammatory and allergic diseases of the eye, including allergic conjunctivitis. The compounds are also useful as cytoprotective agents and for the treatment of migraine headache.

Thus, the compounds of the present invention may also be used to treat or prevent mammalian (especially, human) disease states such as erosive gastritis; erosive esophagitis; inflammatory bowel disease; ethanol-induced hemorrhagic erosions; hepatic ischemia; noxious agent-induced damage or necrosis of hepatic, pancreatic, renal, or myocardial tissue; liver parenchymal damage caused by hepatoxic agents such as CCl₄ and D-galactosamine; ischemic renal failure; disease-induced hepatic damage; bile salt induced pancreatic or gastric damage; trauma- or stress-induced cell damage; and glycerol-induced renal failure.

The compounds of this invention are inhibitors of the biosynthesis of 5-lipoxygenase metabolites of arachidonic acid, such as 5-HPETE, 5-HETE and the leukotrienes. Leukotrienes B₄, C₄, D₄ and E₄ are known to contribute to various disease conditions such as asthma, psoriasis, pain, ulcers and systemic anaphylaxis. Thus inhibition of the synthesis of such compounds will alleviate these and other leukotriene-related disease states.

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- 4 -

DETAILED DESCRIPTION OF THE INVENTION

The present invention provides novel compounds of the formula I:

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Het
$$-X^4$$
 R^5 $(CR^{11}R^{11})_{n}-Y_{m}-(CR^{11-11})_{p}-Q$

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I

wherein:

Het is ArR1R2;

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- Ar is a monocyclic aromatic 5- or 6-member ring containing 1 to 3 N atoms, and the N-oxides thereof;
- 20 R¹, R², R³, R⁴ and R¹⁰ are independently hydr gen, halogen, perhalo lower alkenyl, lower alkyl, lower alkenyl, lower alkynyl, -CF₃, CN, -NO₂, -N₃, -C(OH)R¹¹R¹¹, -CO₂R¹², -14, -S(0)R¹⁴, -S(0)₂R¹⁴, -S(0)₂NR¹⁵R¹⁵, OR¹⁵, OR¹⁵, or -(CH₂)_tR²¹;

 R^5 is hydrogen, $-CH_3$, CF_3 , -C(0)H, $X^1-R^6 = X^2-R^7$;

30 R^6 and R^9 are independently alkyl, alkenyl, $-(CH_2)_u Ph(R^{10})_2$ or $-(CH_2)_u Th(R^{10})_2$

 R^7 is $-C\dot{F}_3$ or R^6 ;

 R^8 is hydrogen or X^3-R^9 ;

- each R¹¹ is independently hydrogen or lower alkyl, or two R¹¹'s on same carbon atom are joined to form a cycloalkyl ring of 3 to 6 carbon atoms;
- 10 R^{12} is hydrogen, lower alkyl or $-CH_2R^{21}$;
 - R^{13} is lower alkyl or $-(CH_2)_rR^{21}$;
 - R^{14} is $-CF_3$ or R^{13} ;

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- hydrogen, -COR¹⁶, R¹³, or two R¹⁵'s on the same nitrogen may be joined to form a monocyclic heterocyclic ring of 4 to 6 atoms containing up to 2 heteroatoms chosen from 0, S, or N;
- R¹⁶ is hydrogen, -CF₃, lower alkyl, lower alkenyl,
- lower alkynyl or -(CH₂)_rR²¹;
- 25 R^{17} is $-(CH_2)_s C(R^{18}R^{18}) (CH_2)_s R^{19}$ or $-CH_2CONR^{15}R^{15}$;
 - R¹⁸ is hydrogen or lower alkyl;

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R<sup>19</sup> is a) a monocyclic or bicyclic heterocyclic ring containing from 3 to 9 nuclear carbon atoms and 1 or 2 nuclear hetero-atoms selected from N, S or 0 and with each ring in the heterocyclic radical being formed of 5 or 6 atoms, or b) the radical W-R<sup>20</sup>;

R<sup>20</sup> is alkyl or -COR<sup>23</sup>;

R<sup>21</sup> is phenyl substituted with 1 or 2 R<sup>22</sup> groups;
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R²² is hydrogen, halogen, lower alkyl, lower alkoxy, lower alkylthio, lower alkylsulfonyl, lower alkylcarbonyl, -CF₃, -CN, -NO₂ or -N₃;

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 R^{24} is the residual structure of a standard amino acid, or R^{18} and R^{24} attached to the same N can cyclize to form a proline residue;

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     m is
               0 or 1;
               0 to 3;
     n is
               1 to 3 when m is 1;
     p is
               0 to 3 when m is 0;
     p is
     r is
               0 to 2;
30
     s is
               0 to 3;
     t is
               0 to 2;
     u is
               0.to 3;
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0, S or NR<sup>15</sup>;
       Wis
                     0 or NR<sup>15</sup>;
       x^1 is
                   .CO, CR^{11}R^{11}, S, S(0), or S(0)<sub>2</sub>;
       x^2 is
                     co, CR^{11}R^{11}, S(0)_2, or a bond;
       x^3 is
                     CH=CH, CH_2-Y^1, or Y^1-CH_2;
       x^4 is
                     x^1 or x^2;
       Y is
       Y^1 is
                     0, S, S(0)_2, or CH_2;
                     -\text{CO}_2\text{R}^{12}, -\text{CONHS}(0)_2\text{R}^{14}, -\text{NHS}(0)_2\text{R}^{14},
       Q is
                     -S(0)_2NHR^{15}, -CONR^{15}R^{15}, -CO_2R^{17},
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                     -CONR<sup>18</sup>R<sup>24</sup>, -CR<sup>11</sup>R<sup>11</sup>OH, or 1H- or
                     2H-tetrazo1-5-y1;
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or a pharmaceutically acceptable salt thereof.

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Y¹ is 0;

A preferred embodiment of Formula I is that in which X^4 is CH_2-Y^1 , Y^1 is 0, and the remaining sustituents are as defined for Formula I.

Another preferred embodiment of Formula I is that in which R1, R2, R3, and R4 are hydrogen; R5 is X2-R7; R7 is R6; R8 is R9; R10 is hydrogen or halogen; m is 0; n is 1 to 3; u is 0 in R6 and 1 in R9; X2 is CR11R11 or S; X4 is CH2-Y1;

 \prod

Q is $-CO_2R^{12}$ or 1-H or 2H-tetrazo1-5-y1; and the remaining substituents are as defined for Formula I; or a pharmaceutically acceptable salt thereof.

5 <u>Definitions</u>

The following abbreviations have the indicated meanings:

10 Me = methy1Bn = benzylPh = phenyl DIBAL-N = diisobutyl alumnium hydride HMPA = hexamethylphosphorictriamide 15 KHMDS = potassium hexamethyldisilazide t-Bu = tert-buty1 i-Pr = isopropy1 $c-C_6H_{11} = cyclohexy1$ c-Pr = cyclopropy120 c- = cycloAc = acety1Tz = 1H- or 2H- tetrazol-5-y1Th = 2- or 3- thieny1 $c-C_5H_9 = cyclopentyl$ 25 1-Ad = 1-adamanty1 NBS = N-bromosuccinimide NCS = N-chlorosuccinimide

Alkyl, alkenyl, and alkynyl are intended to include linear, branched, and cyclic structures and combinations thereof.

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"Alkyl" includes "lower alkyl" and extends to cover carbon fragments having up to 20 carbon atoms. Examples of alkyl groups include octyl, nonyl, norbornyl, undecyl, dodecyl, tridecyl, tetradecyl, pentadecyl, eicosyl, 3,7-diethyl-2,2-dimethyl-4-propylnonyl, cyclododecyl, adamantyl, and the like.

"Lower alkyl" means alkyl groups of from 1 to 7 carbon atoms. Examples of lower alkyl groups include methyl, ethyl, propyl, isopropyl, butyl, secand tert-butyl, pentyl, hexyl, heptyl, cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, 2-methylcyclopropyl, cyclopropylmethyl, and the like.

"Cycloalkyl" refers to a hydrocarbon ring
having from 3 to 7 carbon atoms. Examples of
cycloalkyl groups are cyclopropyl, cyclopentyl,
cycloheptyl, and the like.

"Lower alkenyl" means alkenyl groups of 2 to 7 carbon atoms. Examples of lower alkenyl groups include vinyl, allyl, isopropenyl, pentenyl, hexenyl, heptenyl, cyclopropenyl, cyclobutenyl, cyclopentenyl, cyclohexenyl, 1-propenyl, 2-butenyl, 2-methyl-2-butenyl, and the like.

"Lower alkynyl" means alkynyl groups of 2 to 7 carbon atoms. Examples of lower alkynyl groups include ethynyl, propargyl, 3-methyl-1-pentynyl, 2-heptynyl, and the like.

"Lower alkoxy" means alkoxy groups of from 1
to 7 carbon atoms of a straight, branched, or cyclic

configuration. Examples of lower alkoxy groups
include methoxy, ethoxy, propoxy, isopropoxy,
cyclopropyloxy, cyclohexyloxy, and the like.

- 10 -

"Lower alkylthio" means alkylthio groups of from 1 to 7 carbon atoms of a straight, branched or cyclic configuration. Examples of lower alkylthio groups include methylthio, propylthio, isopropylthio, cycloheptylthio, etc. By way of illustration, the propylthio group signifies -SCH2CH2CH3.

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The term "monocyclic monoheterocyclic ring" which defines R²³ means monocyclic groups of 5 to 7 members containing only 1 heteroatom selected from N, S or 0 in the ring. Examples include tetrahydrofuran, tetrahydrothiophene, pyrrolidine, piperidine, tetrahydropyran, and the like.

The term "monocyclic or bicyclic heterocyclic ring" which defines R¹⁹ may be

2,5-dioxo-1-pyrrolidinyl, (3-pyridinylcarbonyl) amino, 1,3-dihydro-1,3-dioxo-2H-isoindol-2-yl,
1,3-dihydro-2H-isoindol-2-yl, 2,4-imidazolinedion-1-yl, 2,6-piperidinedion-1-yl, 2-imidazolyl,
2-oxo-1,3-dioxolen-4-yl, piperidin-1-yl,
morpholin-1-yl, piperazin-1-yl, and the like.

"Monocyclic aromatic 5- or 6-membered ring containing 1 to 3 N atoms, and the N-oxides thereof" which defines "Ar" may include pyrrole, imidazole, 1,2,3-triazole, 1,2,4-triazole, pyridine, pyridazine, pyrimidine, pyrazine, 1,2,3-, 1,2,4-, or 1,3,5-triazine, and the like.

The point of attachment of any heterocyclic ring may be at any free valence of the ring.

The term standard amino acid is employed to include the following amino acids: alanine, asparagine, aspartic acid, arginine, cysteine, glutamic acid, glutamine, glycine, histidine,

- 11 -

isoleucine, leucine, lysine, methionine, phenylalanine, proline, serine, threonine, tryptophan, tyrosine and valine. (See F.H.C. Crick, Symposium of the Society for Experimental Biology, 1958 (12) p. 140.)

It is understood that \mathbb{R}^1 and \mathbb{R}^2 may be located at any free positions of Ar.

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The terms $Ph(R^{10})_2$ and $Th(R^{10})_2$ indicate a phenyl or thienyl group substituted with two R^{10} substituents.

Halogen includes F, Cl, Br, and I.

It is intended that the definitions of any substituent (e.g., R^1 , R^2 , R^{15} , $Ph(R^{10})_2$, etc.) in a particular molecule be independent of its definitions elsewhere in the molecule. Thus, $-NR^{15}R^{15}$ represents -NHH, $-NHCH_3$, $-NHC_6H_5$, etc.

The monocyclic heterocyclic rings formed when two R¹⁵ groups join through N include pyrrolidine, piperidine, morpholine, thiamorpholine, piperazine, and N-methylpiperazine.

The prodrug esters of Q (i.e., when Q = CO_2R^{17}) are intended to include the esters such as are described by Saari et al., J. Med. Chem., 21, No. 8, 746-753 (1978), Sakamoto et al., Chem. Pharm.

Bull., 32, No. 6, 2241-2248 (1984) and Bundgaard et al., J. Med. Chem., 30, No. 3, 451-454 (1987).

Some of the compounds described herein contain one or more asymmetric centers and may thus give rise to diastereomers and optical isomers. The present invention is meant to comprehend such possible diastereomers as well as their racemic and resolved, enantiomerically pure forms and pharmaceutically acceptable salts thereof.

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The pharmaceutical compositions of the present invention comprise a compound of Formula I as an active ingredient or a pharmaceutically acceptable salt, thereof, and may also contain a 5 pharmaceutically acceptable carrier and optionally other therapeutic ingredients. The term "pharmaceutically acceptable salts" refers to salts prepared from pharmaceutically acceptable non-toxic bases including inorganic bases and organic bases. 10 Salts derived from inorganic bases include aluminum, ammonium, calcium, copper, ferric, ferrous, lithium, magnesium, manganic salts, manganous, potassium, sodium, zinc and the like. Particularly preferred are the ammonium, calcium, magnesium, potassium 15 and sodium salts. Salts derived from pharmaceutically acceptable organic non-toxic bases include salts of primary, secondary, and tertiary amines, substituted amines including naturally occurring substituted amines, cyclic amines and basic ion exchange resins, 20 such as arginine, betaine, caffeine, choline, N, N1-dibenzylethylenediamine, diethylamine, 2-diethylaminoethanol, 2-dimethylaminoethanol, ethanolamine, ethylenediamine, N-ethylmorpholine, N-ethylpiperidine, glucamine, glucosamine, histidine, 25 hydrabamine, isopropylamine, lysine, methylglucamine, morpholine, piperazine, piperidine, polyamine resins, procaine, purines, theobromine, triethylamine, trimethylamine, tripropylamine, tromethamine and the like.

When the compound of the present invention is basic, salts may be prepared from pharmaceutically acceptable non-toxic acids, including inorganic and

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organic acids. Such acids include acetic, benzenesulfonic, benzoic, camphorsulfonic, citric, ethanesulfonic, fumaric, gluconic, glutamic, hydrobromic, hydrochloric, isethionic, lactic, maleic, malic, mandelic, methanesulfonic, mucic, nitric, pamoic, pantothenic, phosphoric, succinic, sulfuric, tartaric, p-toluenesulfonic acid and the like. Particularly preferred are citric, hydrobromic, hydrochloric, maleic, phosphoric, sulfuric and tartaric acids.

It will be understood that in the discussion of methods of treatment which follows, references to the compounds of Formula I are meant to also include the pharmaceutically acceptable salts.

15 The ability of the compounds of Formula I to inhibit biosynthesis of the leukotrienes makes them useful for inhibiting the symptoms induced by the leukotrienes in a human subject. This inhibition of the mammalian biosynthesis of leukotrienes indicates 20 that the compounds and pharmaceutical compositions thereof are useful to treat, prevent or ameliorate in mammals and especially in humans: 1) pulmonary conditions including diseases such as asthma, 2) allergies and allergic reactions such as allergic 25 rhinitis, contact dermatitis, allergic conjunctivitis, and the like, 3) inflammation such as arthritis or inflammatory bowel disease, 4) pain, 5) skin conditions such as psoriasis and the like, 6) cardiovascular conditions such as angina, endotoxin --- 30 shock, and the like and 7) renal insufficiency arising from ischaemia induced by immunological or chemical (cyclosporin) etiology, and that the compounds are cytoprotective agents.

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The cytoprotective activity of a compound may be observed in both animals and man by noting the increased resistance of the gastrointestinal mucosa to the noxious effects of strong irritants, for example, the ulcerogenic effects of aspirin or indomethacin. In addition to lessening the effect of non-steroidal anti-inflammatory drugs on the gastrointestinal tract, animal studies show that cytoprotective compounds will prevent gastric lesions induced by oral administration of strong acids, strong bases, ethanol, hypertonic saline solutions and the like.

Two assays can be used to measure cytoprotective ability. These assays are; (A) an ethanol-induced lesion assay and (B) an indomethacin-induced ulcer assay and are described in EP 140.684.

The magnitude of prophylactic or therapeutic dose of a compound of Formula I will, of course, vary with the nature of the severity of the condition to be treated and with the particular compound of Formula I and its route of administration. also vary according to the age, weight and response of the individual patient. In general, the daily dose range for anti-asthmatic, anti-allergic or anti-inflammatory use and generally, uses other than cytoprotection, lie within the range of from about 0.001 mg to about 100 mg per kg body weight of a mammal, preferably 0.01 mg to about 10 mg per kg, and most preferably 0.1 to 1 mg per kg, in single or divided doses. On the other hand, it may be necessary to use dosages outside these limits in some cases.

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For use where a composition for intravenous administration is employed, a suitable dosage range for anti-asthmatic, anti-inflammatory or anti-allergic use is from about 0.001 mg to about 25 mg (preferably from 0.01 mg to about 1 mg) of a compound of Formula I per kg of body weight per day and for cytoprotective use from about 0.1 mg to about 100 mg (preferably from about 1 mg to about 100 mg and more preferably from about 1 mg to about 10 mg) of a compound of Formula I per kg of body weight per day.

In the case where an oral composition is employed, a suitable dosage range for anti-asthmatic, anti-inflammatory or anti-allergic use is, e.g. from about 0.01 mg to about 100 mg of a compound of Formula I per kg of body weight per day, preferably from about 0.1 mg to about 10 mg per kg and for cytoprotective use from 0.1 mg to about 100 mg (preferably from about 1 mg to about 100 mg and more preferably from about 10 mg to about 100 mg) of a compound of Formula I per kg of body weight per day.

For the treatment of diseases of the eye, ophthalmic preparations for ocular administration comprising 0.001-1% by weight solutions or suspensions of the compounds of Formula I in an acceptable ophthalmic formulation may be used.

The exact amount of a compound of the Formula I to be used as a cytoprotective agent will depend on, inter alia, whether it is being administered to heal damaged cells or to avoid future damage, on the nature of the damaged cells (e.g., gastrointestinal ulcerations vs. nephrotic necrosis),

and on the nature of the causative agent. An example of the use of a compound of the Formula I in avoiding future damage would be co-administration of a compound of the Formula I with a non-steroidal anti-inflammatory drug (NSAID) that might otherwise cause such damage (for example, indomethacin). For such use, the compound of Formula I is administered from 30 minutes prior up to 30 minutes after administration of the NSAID. Preferably it is administered prior to or simultaneously with the NSAID, (for example, in a combination dosage form).

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Any suitable route of administration may be employed for providing a mammal, especially a human with an effective dosage of a compound of the present invention. For example, oral, rectal, topical, parenteral, ocular, pulmonary, nasal, and the like may be employed. Dosage forms include tablets, troches, dispersions, suspensions, solutions, capsules, creams, ointments, aerosols, and the like.

The pharmaceutical compositions of the present invention comprise a compound of Formula I as an active ingredient or a pharmaceutically acceptable salt thereof, and may also contain a pharmaceutically acceptable carrier and optionally other therapeutic ingredients. The term "pharmaceutically acceptable salts" refers to salts prepared from pharmaceutically acceptable non-toxic bases or acids including inorganic bases or acids and organic bases or acids.

The compositions include compositions

suitable for oral, rectal, topical, parenteral

(including subcutaneous, intramuscular, and
intravenous), ocular (ophthalmic), pulmonary (nasal

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or buccal inhalation), or nasal administration, although the most suitable route in any given case will depend on the nature and severity of the conditions being treated and on the nature of the active ingredient. They may be conveniently presented in unit dosage form and prepared by any of the methods well-known in the art of pharmacy.

For administration by inhalation, the compounds of the present invention are conveniently delivered in the form of an aerosol spray presentation from pressurized packs or nebulisers. The compounds may also be delivered as powders which may be formulated and the powder composition may be inhaled with the aid of an insufflation powder inhalation is a metered delivery system for inhalation is a metered dose inhalation (MDI) aerosol, which may be formulated as a suspension or solution of Compound I in suitable propellants, such as fluorocarbons or hydrocarbons.

Suitable topical formulations of Compound I include transdermal devices, aerosols, creams, ointments, lotions, dusting powders, and the like.

In practical use, the compounds of Formula I can be combined as the active ingredient in intimate admixture with a pharmaceutical carrier according to conventional pharmaceutical compounding techniques.

The carrier may take a wide variety of forms depending on the form of preparation desired for administration, e.g., oral or parenteral (including intravenous). In preparing the compositions for oral dosage form, any of the usual pharmaceutical media may be employed, such as, for example, water,

- 18 -

glycols, oils, alcohols, flavoring agents, preservatives, coloring agents and the like in the case of oral liquid preparations, such as, for example, suspensions, elixirs and solutions; or carriers such as starches, sugars, microcrystalline cellulose, diluents, granulating agents, lubricants, binders, disintegrating agents and the like in the case of oral solid preparations such as, for example, powders, capsules and tablets, with the solid oral preparations being preferred over the liquid preparations. Because of their ease of administration, tablets and capsules represent the most advantageous oral dosage unit form in which case solid pharmaceutical carriers are obviously employed. If desired, tablets may be coated by

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In addition to the common dosage forms set out above, the compounds of Formula I may also be administered by controlled release means and/or delivery devices such as those described in U.S. Patent Nos. 3,845,770; 3,916,899; 3,536,809; 3,598,123; 3,630,200 and 4,008,719.

standard aqueous or nonaqueous techniques.

25 presented as discrete units such as capsules, cachets or tablets each containing a predetermined amount of the active ingredient, as a powder or granules or as a solution or a suspension in an aqueous liquid, a non-aqueous liquid, an oil-in-water emulsion or a water-in-oil liquid emulsion. Such compositions may be prepared by any of the methods of pharmacy but all methods include the step of bringing into association

the active ingredient with the carrier which constitutes one or more necessary ingredients. general, the compositions are prepared by uniformly and intimately admixing the active ingredient with liquid carriers or finely divided solid carriers or both, and then, if necessary, shaping the product into the desired presentation. For example, a tablet may be prepared by compression or molding, optionally with one or more accessory ingredients. Compressed 10 tablets may be prepared by compressing in a suitable machine, the active ingredient in a free-flowing form such as powder or granules, optionally mixed with a binder, lubricant, inert diluent, surface active or was active dispersing agent. Molded tablets may be made by 15 molding in a suitable machine, a mixture of the powdered compound moistened with an inert liquid diluent. Desirably, each tablet contains from about 2.5 mg to about 500 mg of the active ingredient and each cachet or capsule contains from about 2.5 to 20 about 500 mg of the active ingredient. The following are examples of representative

pharmaceutical dosage forms for the compounds of Formula I:

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	Injectable Suspension (I.M.)	mg/mL
	Compound of Formula I	10
	Methylcellulose	5.0
5	Tween 80	0.5
	Benzyl alcohol	9.0
	Benzalkonium chloride	1.0
	Water for injection	
	to a total volume of 1 mL	
10		
	<u>Tablet</u>	mg/tablet
	Compound of Formula I	25
•	Microcrystalline Cellulose	415
	Providone	14.0
15	Pregelatinized Starch	43.5
•	Magnesium Stearate	2.5
	·	500
•		
	Capsule	mg/capsule
20	Compound of Formula I	25
	Lactose Powder	573.5
	Magnesium Stearate	1.5
		600
25	Aerosol ·	Per canister
	Compound of Formula I	24 mg
	Lecithin, NF Liquid Concentrate	1.2 mg
	Trichlorofluoromethane, NF	4.025 gm
	Dichlorodifluoromethane, NF	12.15 gm
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- 21 -

In addition to the compounds of Formula I, the pharmaceutical compositions of the present invention can also contain other active ingredients, such as cyclooxygenase inhibitors, non-steroidal anti-inflammatory drugs (NSAIDs), peripheral analgesic agents such as zomepirac diflunisal and the like. The weight ratio of the compound of the Formula I to the second active ingredient may be varied and will depend upon the effective dose of 10 each ingredient. Generally, an effective dose of each will be used. Thus, for example, when a compound of the Formula I is combined with an NSAID the weight ratio of the compound of the Formula I to the NSAID will generally range from about 1000:1 to 15 about 1:1000, preferably about 200:1 to about 1:200. Combinations of a compound of the Formula I and other active ingredients will generally also be within the aforementioned range, but in each case, an effective dose of each active ingredient should be used.

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20 NSAIDs can be characterized into five groups:

- (1) the propionic acid derivatives;
- (2) the acetic acid derivatives;
- the fenamic acid derivatives; (3)
- (4) the oxicams; and

25 (5) the biphenylcarboxylic acid derivatives:

or a pharmaceutically acceptable salt thereof.

The propionic acid derivatives which may be used comprise: alminoprofen, benoxaprofen, bucloxic acid, carprofen, fenbufen, fenoprofen, fluprofen, flurbiprofen, ibuprofen, indoprofen, ketoprofen,

- 22 -

miroprofen, naproxen, oxaprozin, pirprofen, prano-profen, suprofen, tiaprofenic acid, and tioxaprofen. Structurally related propionic acid derivatives having similar analgesic and anti-inflammatory properties are also intended to be included in this group.

Thus, "propionic acid derivatives" as

defined herein are non-narcotic

analgesics/non-steroidal anti-inflammatory drugs

having a free -CH(CH₃)COOH or -CH₂CH₂COOH group

(which optionally can be in the form of a

pharmaceutically acceptable salt group, e.g.,

-CH(CH₃)COO-Na+ or -CH₂CH₂COO-Na+),

typically attached directly or via a carbonyl

function to a ring system, preferably to an aromatic
ring system.

The acetic acid derivatives which may be used comprise: indomethacin, which is a preferred NSAID, acemetacin, alclofenac, clidanac, diclofenac, fenclofenac, fenclozic acid, fentiazac, furofenac, ibufenac, isoxepac, oxpinac, sulindac, tiopinac, tolmetin, zidometacin and zomepirac. Structurally related acetic acid derivatives having similar analgesic and anti-inflammatory properties are also intended to be encompassed by this group.

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Thus, "acetic acid derivatives" as defined herein are non-narcotic analgesics/non-steroidal anti-inflammatory drugs having a free -CH2COOH group (which optionally can be in the form of a pharmaceutically acceptable salt group, e.g. -CH2COO-Na+), typically attached directly to a ring system, preferably to an aromatic or heteroaromatic ring system.

The fenamic acid derivatives which may be used comprise: flufenamic acid, meclofenamic acid, mefenamic acid, niflumic acid and tolfenamic acid. Structurally related fenamic acid derivatives having similar analgesic and anti-inflammatory properties are also intended to be encompassed by this group.

Thus, "fenamic acid derivatives" as defined herein are non-narcotic analgesics/non-steroidal anti-inflammatory drugs which contain the basic structure:

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which can bear a variety of substituents and in which the free -COOH group can be in the form of a pharmaceutically acceptable salt group, e.g., -COO-Na+.

The biphenylcarboxylic acid derivatives which can be used comprise: diflunisal and flufenisal. Structurally related biphenylcarboxylic acid derivatives having similar analgesic and anti-inflammatory properties are also intended to be encompassed by this group.

Thus, "biphenylcarboxylic acid derivatives"
as defined herein are non-narcotic
analgesics/non-steroidal anti-inflammatory drugs
which contain the basic structure:

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which can bear a variety of substituents and in which the free -COOH group can be in the form of a pharmaceutically acceptable salt group, e.g., -COO-Na+.

The oxicams which can be used in the present invention comprise: isoxicam, piroxicam, sudoxicam and tenoxican. Structurally related oxicams having similar analgesic and anti-inflammatory properties are also intended to be encompassed by this group.

Thus, "oxicams" as defined herein are nonnarcotic analgesics/non-steroidal anti-inflammatory drugs which have the general formula:

wherein R is an aryl or heteroaryl ring system.

The following NSAIDs may also be used:
amfenac sodium, aminoprofen, anitrazafen,
antrafenine, auranofin, bendazac lysinate,
benzydanine, beprozin, broperamole, bufezolac,
cinmetacin, ciproquazone, cloximate, dazidamine,

deboxamet, delmetacin, detomidine, dexindoprofen, diacerein, di-fisalamine, difenpyramide, emorfazone, enfenamic acid, enolicam, epirizole, etersalate, etodolac, etofenamate, fanetizole mesylate,

- fenciorac, fendosal, fenflumizole, feprazone, floctafenine, flunixin, flunoxaprofen, fluproquazone, fopirtoline, fosfosal, furcloprofen, glucametacin, guaimesal, ibuproxam, isofezolac, isonixim, isoprofen, isoxicam, lefetamine HCl, leflunomide,
- lofemizole, lonazolac calcium, lotifazole, loxoprofen, lysin clonixinate, meclofenamate sodium, meseclazone, nabumetone, nictindole, nimesulide, orpanoxin, oxametacin, oxapadol, perisoxal citrate, pimeprofen, pimetacin, piproxen, pirazolac,
- pirfenidone, proglumetacin maleate, proquazone,
 pyridoxiprofen, sudoxicam, talmetacin, talniflumate,
 tenoxicam, thiazolinobutazone, thielavin B, tiaramide
 HC1, tiflamizole, timegadine, tolpadol, tryptamid and
 ufenamate.
- The following NSAIDs, designated by company code number (see e.g., Pharmaprojects), may also be used: 480156S, AA861, AD1590, AFP802, AFP860, AI77B, AP504,
- AU8001, BPPC, BW540C, CHINOIN 127, CN100, EB382,

 EL508, F1044, GV3658, ITF182, KCNTEI6090, KME4,

 LA2851; MR714, MR897, MY309, ON03144, PR823, PV102,

 PV108, R830, RS2131, SCR152, SH440, SIR133, SPAS510,

 SQ27239, ST281, SY6001, TA60, TAI=901 (4-benzoy1-1
 indancarboxylic acid), TVX2706, U60257, UR2301, and

 WV41770

Finally, NSAIDs which may also be used include the salicylates, specifically acetyl salicylic acid

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and the phenylbutazones, and pharmaceutically acceptable salts thereof.

In addition to indomethacin, other preferred NSAIDS are acetyl salicylic acid, diclofenac, fenbufen, fenoprofen, flurbiprofen, ibuprofen, ketoprofen, naproxen, phenylbutazone, piroxicam, sulindac and tolmetin.

Pharmaceutical compositions comprising the Formula I compounds may also contain inhibitors of the biosynthesis of the leukotrienes such as are disclosed in EP 138,481 (April 24,1985), EP 115,394 (August 8, 1984), EP 136,893 (April 10, 1985), and EP 140,709 (May 8, 1985), which are hereby incorporated herein by reference.

The compounds of the Formula I may also be used in combination with leukotriene antagonists such as those disclosed in EP 106,565 (April 25, 1984) and EP 104,885 (April 4, 1984) which are hereby incorporated herein by reference and others known in the art such as those disclosed in EP Application Nos. 56,172 (July 21, 1982) and 61,800 (June 10, 1982); and in U.K. Patent Specification No. 2,058,785 (April 15, 1981), which are hereby incorporated herein by reference.

Pharmaceutical compositions comprising the Formula I compounds may also contain as the second active ingredient, prostaglandin antagonists such as those disclosed in EP 11,067 (May 28, 1980) or thromboxane antagonists such as those disclosed in U.S. Pat. 4,237,160. They may also contain histidine decarboxylase inhibitors such as α -fluoromethylhistidine, described in U.S. Pat. 4,325,961. The

compounds of the Formula I may also be advantageously

combined with an H₁ or H₂-receptor antagonist, such as for instance acetamazole, aminothiadiazoles disclosed in EP 40,696 (December 2, 1981), benadry1, 5 cimetidine, famotidine, framamine, histadyl, phenergan, ranitidine, terfenadine and like compounds, such as those disclosed in U.S. Patent Nos. 4,283,408; 4,362,736; and 4,394,508. pharmaceutical compositions may also contain a K+/H+ 10 ATPase inhibitor such as omeprazole, disclosed in U.S. Pat. 4,255,431, and the like. Compounds of Formula I may also be usefully combined with most cell stabilizing agents, such as 1,3-bis(2-carboxychromon-5-yloxy)-2-hydroxypropane and related 15 compounds described in British Patent Specifications 1,144,905-and-1,144,906. Another-useful pharmaceutical composition comprises the Formula I compounds in combination with serotonin antagonists such as methysergide, the serotonin antagonists 20 described in Nature, Vol. 316, pages 126-131, 1985, and the like. Each of the references referred to in this paragraph is hereby incorporated herein by reference. Other advantageous pharmaceutical 25 compositions—comprise—the Formula I compounds in

compositions comprise the Formula I compounds in combination with anti-cholinergics such as ipratropium bromide, bronchodilators such as the beta agonist salbutamol, metaproterenol, terbutaline, fenoterol and the like, and the anti-asthmatic drugs theophylline, choline theophyllinate and enprofylline, the calcium antagonists nifedipine, diltiazem, nitrendipine, verapamil, nimodipine,

felodipine, etc. and the corticosteroids, hydrocortisone, methylprednisolone, betamethasone, dexamethasone, beclomethasone, and the like.

Compounds of the present invention can be prepared according to the following methods.

Temperatures are in degrees Celsius.

The starting methoxy phenylhydrazines <u>II</u> are either commercially available or are described in the chemical literature as are the acetamidophenols <u>XXVI</u>. The benzyl phenylhydrazine starting materials <u>III</u> are prepared as described in EP 166,591 (17102 IA) and the ketones <u>IV</u> and <u>XXXI</u> are prepared as described in EP 166,591 and EP 275,667 (17496 IA).

- The 2-(halomethyl)quinolines <u>VII</u> are available from literature methods described in "Quinolines" Parts I and II, G. Jones (ED.), John Wiley & Sons, Toronto, 1977 and 1982. The preparation of <u>VII</u> by halogenation of the corresponding 2-methylquinolines is also described in the Jones' volumes. The benzyl halides,
 - described in the Jones' volumes. The benzyl halides, $(R^{10})_2$ PhCH₂-Hal, are readily prepared and many such compounds are described in the prior art, such as U.S. Patent 4,808,608 (17323 IB). Hal in <u>VII</u> and $(R^{10})_2$ PhCH₂-Hal represents C1, Br or I.

Many syntheses of indoles are well-known in the chemical literature: see for example, "Heterocyclic compounds" Volume 25, Parts I, II, III, W.J. Houlihan (Ed.), Interscience, J. Wiley & Sons, N.Y., 1979, and "The Chemistry of Indoles" by R.J. Sundberg, Academic Press, N.Y., 1970. One of the most common syntheses is known as the Fischer Indole

most common syntheses is known as the Fischer Indole Synthesis, and is abbreviated in the following methods as "Fischer".

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The $-\text{CO}_2\text{H}$ and $-\text{CO}_2\text{R}^{12}$ groups in the intermediates and final products in the various methods can be transformed to other representatives of Q such as $-\text{CONHS}(0)_2\text{R}^{14}$, $-\text{NHS}(0)_2\text{R}^{14}$, $-\text{CONR}^{15}\text{R}^{15}$, $-\text{CH}_2\text{OH}$ or tetrazol-5-yl by the methodology described in U.S. Patent 4,808,608 (17323IB). The preparation of the pro-drug forms (Q is $-\text{CO}_2\text{R}^{17}$) from the acids may be effected by the methodology of EP 104,885 (16830 IA).

It will be apparent to one skilled in the art that the various functional groups (R¹, R², Y, Q, etc.) must be chosen so as to be compatible with the chemistry being carried out. Such compatibility can often be achieved by protecting groups, or by specific variations in the sequence of the reactions.

When R⁵ is S-R⁷, the corresponding sulfoxides and sulfones can be prepared by oxidation of the sulfides with one or two equivalents of an oxidizing agent such as m-chloroperbenzoic acid or monoperoxyphthalic acid or oxone (Trost, J. Org. Chem., 1988, pg.532).

Many of the following methods involve a basic hydrolysis of an ester function to obtain the corresponding carboxylic acid. In all cases, the free acid is obtained by acidification of the reaction mixture with a suitable acid such as hydrochloric, sulfuric, acetic, trifluoroacetic acid, etc.

Compounds 6, 10, 11, 16, 17, 19, 23, 24, 27,

^{28,} and their precursor esters are all examples of the Formula I compounds of the present invention.

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Compounds identified by Roman numerals (IV, V, XIV, XXVI, XXXI, and XXXV) are known and correspond to those compounds in EP 419,049, which is incorporated herein by reference.

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WO 94/13662

- 31 -

METHOD 1

hydride reducing

Het-
$$CO_2R^{12}$$

Agent

 $\frac{1}{2}$

Het- CH_2OH

Het- CH_2LG

METHOD 2

base

Het-
$$CH_2O$$
 R^4
 R^5
 CO_2Me

$$\frac{6}{R^8}$$

- 32 -

Method 1

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The carboxy derivative 1 may be reduced by a suitable hydride reducing agent such as lithium aluminum hydride, sodium borohydride, DIBAL-H, or the like in appropriate solvents such as ether, THF, hexane, toluene, or mixtures thereof, to obtain alcohol 2. The alcohol function of 2 can be converted to a suitable leaving group (LG) such as a halide, or a sulfonate ester (mesylate, tosylate, triflate, etc.) by methods well known in the art to produce intermediate 3. A useful subgroup of 3 can be prepared by halogenation of the methyl compound Het-CH3 by heating with halogenating agents such as NCS or NBS in appropriate solvents such as carbon tetrachloride, benzene, and the like.

Reaction of 3 with triphenylphosphine in ether, acetonitrile, THF, or similar solvents produces the phosphonium salt 4. Compound 4 is converted into the ylid 5 by treating with a base such as Et₃N, sodium hydride, butyl lithium, or an alkowide, depending upon the reactivity of the phosphonium salt 4.

Method 2

Compound 3 is reacted with phenol XIV, in the presence of a suitable base such as potassium or cesium carbonate in a suitable solvent such as acetone, acetonitrile, or DMF to yield compound 6 which can be converted to its corresponding carboxylic acid by standard procedures.

- 33 -

METHOD 3

5 HO NHAC
$$\frac{K_2CO_3}{DMF}$$
 Het- CH_2O NHAC $\frac{XXVI}{3}$ $\frac{3}{10}$ $\frac{7}{KOH}$ $\frac{K_2CO_3}{Het-CH_2O}$ Het- CH_2O NHAC $\frac{R^4}{H_2O/EtOH}$ heat $\frac{R^4}{H_2O/EtOH}$ heat $\frac{R^4}{H_2O/EtOH}$ $\frac{R^4}{H_2O/$

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- 34 -

Method 3

A suitable N-acetylated aminophenol XXVI is reacted with 3 using an alkali hydride or carbonate, such as potassium carbonate as a base in a polar solvent like DMF or NMP. The resulting acetanilide 7 is then de-acetylated using standard basic conditions, preferably using alcoholic potassium hydroxide under reflux to produce the aniline derivative 8. Conversion of the aniline derivative to the hydrazine analogue 9 is effected through reduction of the intermediate diazonium salt using sodium hydrosulfite in an aqueous medium.

The hydrazine 9 is then processed using a Fischer indolization with ketone IV to produce compound 10, which is then alkylated on the indole nitrogen using R⁸-Hal and a suitable base such as KHMDS in THF or NaH in DMF to give compound 11.

. The ...

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METHOD 4

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Het
$$R^3$$
 R^3 R^{11} R^{11} R^{11} R^3 R^3 R^4 R^5 R^5

WO 94/13662 PCT/CA92/00547

- 36 -

Method 4

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Indole phenol <u>XIV</u> is transformed to a phenol triflate 12 by treatment with trifluoromethyl sulfonic anhydride (Tf20) in a solvent like pyridine in dichloromethane. The phenol triflate may be carboxymethylated to a compound like 13 under palladium acetate catalysis in an atmosphere of carbon monoxide, a phosphine ligand like 1,1-bis(diphenylphosphinoferrocene) enhances this reaction. 10 Reduction of the carboxymethylated indole may be effected with a variety of hydride reducing agents. Conveniently, DIBAL-H is used in THF on the hydrolysed ester. The reduced carbinol product 14 is conveniently oxidized to a formylated derivative 15 15 with manganese dioxide in methylene chloride as a typical solvent. Aldehyde 15 can then be homologated under carbanion conditions, typically using Wittig reagent 5 as shown in the method, under anydrous conditions in an etherial solvent like THF. 20 temperature of this reaction is typically from -70°C to room temperature. Indole styryl analogues (trans) 16 are thus formed. Further transformation of the styryl system may be effected by catalytic reduction using H2 and Pd/C in an organic solvent like ethyl 25 acetate to yield the saturated compound 17.

METHOD 5

- 38 -

Method 5

Indole thio analogues of I such as 23 and 24 are conveniently prepared by the sequence shown in Method 5. The treatment of compound \underline{V} with BBr₃ in a 5 chlorinated solvent such as CH2Cl2 cleaves both the methyl ether and the indole N-benzyl group and cyclizes the product to an indole lactam 18. Derivatization of this compound as an N,N-dimethylthiocarbamoyl indole 19 followed by thermal 10 rearrangement at >200°C gives rise to an N,N-dimethy1carbamoylthioindole derivative 20. Depending on the duration of heating, dethiolation $(R^5 = -S - t - Bu \rightarrow R^5 = H)$ may also take place. The hydrolysis of 20 may be effected using strong base, typically sodium 15 methoxide in methanol is used. Spontaneous formation of disulfide 21 may occur in this reaction. reduction of 21 can be achieved using tripheny1phosphine in ageuous dioxane to produce 22. Coupling of 22 to an appropriately substituted derivative 3 20 takes place under organic base catalysis. Typically triethylamine, in an organic solvent such as methylene chloride, is used. Transformation of indole 23 to an N-substituted derivative 24 is achieved under standard conditions described in 25 Method 3.

- 39 -

METHOD 6

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Method 6

Hydrazine 9 may also be transformed directly to unsubstituted indoles by a Fischer reaction with various ketones like XXXI. N-Alkylation of the indoles is effected using the conditions described in Method 3 to produce hetmethoxyindole alkanoate esters 25. Such esters are transformed to ketones or carbinols via Grignard conditions using alkyl magnesium halides in ether solvents like diethyl ether or through the use of lithium aluminum hydride in ether solvents like THF. The carbinols 27 so produced may be further transformed into ester compounds of the present invention by reacting with halo esters XXXV using sodium hydride as base in a suitable solvent like THF. Subsequent hydrolysis of the esters leads to acid compounds 28 of the present invention.

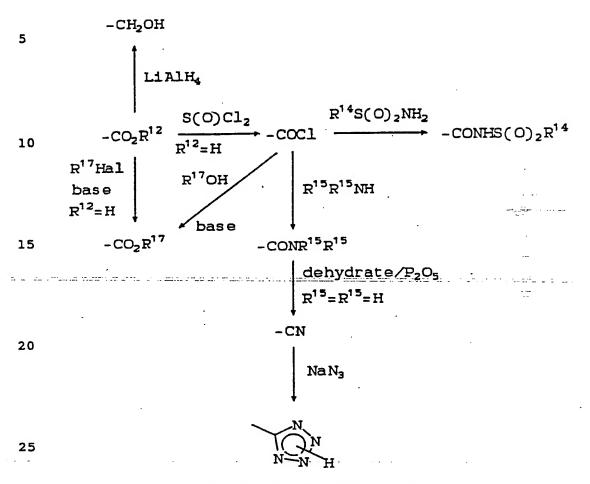
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WO 94/13662 PCT/CA92/00547

- 41 -

METHOD 7



1H- or 2H-tetrazol-5-yl

Method 7

The preparation of the various definitions of Q is outlined in Method 7, starting from the readily available carboxylic acid derivative $-\text{CO}_2\text{R}^{12}$.

- It will be obvious to one skilled in the art that many of the reactions indicated are reversible. Thus, by way of illustration, the -CN group can serve as the starting material to prepare the amide and carboxylic acid functional groups. The reactions depicted in
- Method 7, as well as methods for synthesis of the sulfonamide group (-S(0)₂NHR¹⁵), are well-known in the art. See, for instance, the following textbooks:
- J. March, <u>Advanced Organic Chemistry</u>, 3rd ed., J.
 Wiley and Sons, Toronto, 1985;
 - S.R. Sandler and W. Karo, <u>Organic Functional Group Preparations</u>. I & II, Academic Press, Toronto, 1983 and 1986.

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Representative Compounds

Table I illustrates compounds representative of the present invention.

WO 94/13662 PCT/CA92/00547

- 43 -

· TABLE I

WO 94/13662 PCT/CA92/00547

- 44 -

Assavs for Determining Biological Activity

Compounds of Formula I can be tested using the following assays to determine their mammalian leukotriene biosynthesis inhibiting activity.

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Rat Peritoneal Polymorphonuclear (PMN) Leukocyte Assay Rats under ether anesthesia are injected (i.p.) with 8 mL of a suspension of sodium caseinate 10 (6 grams in ca. 50 mL water). After 15-24 hr. the rats are sacrificed (CO₂) and the cells from the peritoneal cavity are recovered by lavage with 20 mL of buffer (Eagles MEM containing 30 mM HEPES adjusted to pH 7.4 with NaOH). The cells are pelleted (350 \times 15 g, 5 min.), resuspended in buffer with vigorous shaking, filtered through lens paper, recentrifuged and finally suspended in buffer at a concentration of 10 cells/mL. A 500 mL aliquot of PMN suspension and test compound are preincubated for 2 minutes at 37°C. 20 followed by the addition of 10 mM A-23187. suspension is stirred for an additional 4 minutes then bioassayed for LTB4 content by adding an aliquot to a second 500 mL portion of the PMN at 37°C. LTB4 produced in the first incubation causes 25 aggregation of the second PMN, which is measured as a change in light transmission. The size of the assay aliquot is chosen to give a submaximal transmission change (usually -70%) for the untreated control. percentage inhibition of LTB4 formation is calculated 30 from the ratio of transmission change in the sample to the transmission change in the compound-free control.

WO 94/13662 PCT/CA92/00547

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- 45 -

Human Polymorphonuclear (PMN) Leukocyte LTB4 Assay Preparation of Human PMN. Human blood is obtained by antecubital venepuncture from consenting volunteers who have not taken medication within the previous 7 days. The blood is immediately added to 10% (v/v) trisodium citrate (0.13 M) or 5% (v/v) sodium heparin (1000 IU/mL). PMNs are isolated from anticoagulated blood by dextran sedimentation of erythrocytes followed by centrifugation through Ficoll-Hypaque (specific gravity 1.077), as described by Boyum (Scand. J. Clin. Lab. Invest., 21 (Supp. 97), 77(1968)). Contaminating erythrocytes are removed by lysis following exposure to ammonium chloride (0.16 M) in Tris buffer (pH 7.65), and the PMNs resuspended at 5×10^5 cells/mL in HEPES (15) mM)-buffered-Hanks balanced salt solution containing Ca^{2+} (1.4 mM) and Mg^{2+} (0.7 mM), pH 7.4. Viability

Generation and Radioimmunoassay of LTB4. 20 PMNs (0.5 mL; 2.5 x 10^5 cells) are placed in plastic tubes and incubated (37°C, 2 min) with test compounds at the desired concentration or vehicle (DMSO, final concentration 0.2%) as control. The synthesis of 25 LTB4 is initiated by the addition of calcium ionophore A23187 (final concentration 10 mM) or vehicle in control samples and allowed to proceed for -5 minutes at -37°C. The reactions are then terminated by the addition of cold methanol (0.25 mL) and samples of the entire PMN reaction mixture removed 30

is assessed by Trypan blue exclusion.

for radioimmunoassay of LTB4.

· Samples (50 mL) of authentic LTB4 of known concentration in radioimmunoassay buffer (RIA) buffer (potassium phosphate 1 mM; disodium EDTA 0.1 mM; Thimerosal 0.025 mM; gelatin 0.1%, pH 7.3) or PMN reaction mixture diluted 1:1 with RIA buffer are added to reaction tubes. Thereafter, [3H]-LTB₄ (10 nCi in 100 mL RIA buffer) and LTB4-antiserum (100 mL of a 1:3000 dilution in RIA buffer) are added and the tubes vortexed. Reactants are allowed to equilibrate 10 by incubation overnight at 4°C. To separate antibody-bound from free LTB4, aliquots (50 mL) of activated charcoal (3% activated charcoal in RIA buffer containing 0.25% Dextran T-70) are added, the tubes vortexed, and allowed to stand at room 15 temperature for 10 minutes prior to centrifugation (1500 x g; 10 min; 4°C). The supernatants containing antibody-bound LTB4 are decanted into vials and Aquasol 2 (4 mL) added. Radioactivity is quantified by liquid scintillation spectrometry. The 20 specificity of the antiserum and the sensitivity of the procedure have been described by Rokach et al. (Prostaglandins Leukotrienes and Medicine, 1984, 13, 21.) The amount of LTB4 produced in test and control (approx. 20 ng/106 cells) samples is calculated. 25 Inhibitory dose-response curves are constructed using a four-parameter algorithm and from these the IC50 values are determined.

Asthmatic Rat Assav

Rats are obtained from an inbred line of asthmatic rats. Both female (190-250 g) and male (260-400 g) rats are used.

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Egg albumin (EA), grade V, crystallized and lyophilized, is obtained from Sigma Chemical Co., St. Louis. Aluminum hydroxide is obtained from the Regis Chemical Company, Chicago. Methysergide bimaleate is supplied by Sandoz Ltd., Basel.

The challenge and subsequent respiratory recordings are carried out in a clear plastic box with internal dimensions 10x6x4 inches. The top of the box is removable; in use, it is held firmly in place by four clamps and an airtight seal is maintained by a soft rubber gasket. Through the center of each end of the chamber a DeVilbiss nebulizer (No. 40) is inserted via an airtight seal and each end of the box also has an outlet. A Fleisch No. 0000 pneumotachograph is inserted into

- one end of the box and coupled to a Grass volumetric pressure transducer (PT5-A) which is then connected to a Beckman Type R Dynograph through appropriate couplers. While aerosolizing the antigen, the
 - 20 outlets are open and the pneumotachograph is isolated from the chamber. The outlets are closed and the pneumotachograph and the chamber are connected during the recording of the respiratory patterns. For challenge, 2 mL of a 3% solution of antigen in saline
- 25 is placed into each nebulizer and the aerosol is generated with air from a small Potter diaphragm pump operating at 10 psi and a flow of 8 liters/minute.

Rats are sensitized by injecting (subcutaneously) 1 mL of a suspension containing 1 mg

EA and 200 mg aluminum hydroxide in saline. They are 30

used between days 12 and 24 postsensitization. In order to eliminate the serotonin component of the response, rats are pretreated intravenously 5 minutes prior to aerosol challenge with 3.0 mgm/kg of methysergide. Rats are then exposed to an aerosol of 3% EA in saline for exactly 1 minute, then their respiratory profiles are recorded for a further 30 minutes. The duration of continuous dyspnea is measured from the respiratory recordings.

10 Compounds are generally administered either orally 1-4 hours prior to challenge or intravenously 2 minutes prior to challenge. They are either dissolved in saline or 1% methocel or suspended in 1% methocel. The volume injected is 1 mL/kg 15 (intravenously) or 10 mL/kg (orally). Prior to oral treatment rats are starved overnight. Their activity is determined in terms of their ability to decrease the duration of symptoms of dyspnea in comparison with a group of vehicle-treated controls. Usually, a 20 compound is evaluated at a series of doses and an ED50 is determined. This is defined as the dose (mg/kg) which would inhibit the duration of symptoms by 50%.

The invention is further defined by reference to the following examples, which are intended to be illustrative and not limiting. All temperatures are in degrees Celsius.

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INTERMEDIATES

Preparation 1: Methyl 3-[1-(4-chlorobenzyl)-3-methyl-5-hydroxy-indo1-2-y1]-2,2-dimethy1propanoate To a solution of 1.05 g (2.7 mmol) of 3-[1-(4-chlorobenzy1)-3-methy1-5-methoxyindo1-2-y1]-2,2-dimethylpropanoic acid (EP 166,591, Example 22) and 800 µL of ethanethiol (10 mmol) in 20 mL of CH₂Cl₂ at -20°C was added in portions 2.17 g (16 mmol) of AlCl3. The reaction turned light orange and was stirred at room temperature overnight. In the morning, the reaction was completed (tlc) and it was poured into a solution of 1N HCl and extracted 3x with CH2Cl2. The combined organic layers were washed. with brine, dried (MgSO4), and filtered. The filtrate was evaporated and to the residual syrup (680 mg) was added 20 mL of Et₂O followed by an ethereal solution of diazomethane. Evaporation of the solvent left the crude title compound which was used without further purification. ¹H NMR (250 MHz, CDC1₃): δ 7.3-7.15 (m, 3H, aromatic); 6.96 (m, 1H, aromatic): 6.70 (m, 3H, aromatic); 5.34

> Preparation 2: Methy1-3-[1-(4-chlorobenzy1)-3-(tbutylthio)-5-hydroxyindo1-2-y1]-

(s, 2H, N-CH₂); 4.8-4.5 (M, 1H, -OH); 3.76 (s, 3H,

 $25 - CO_2Me$; $3.12 (s, 2H, 2-CH_2)$; 2.40 (s, 3H, 3-Me);

30 2.2-dimethylpropanoate

1.44 (s, 6H, $C(Me)_2$).

The title compound was prepared as described in EP 419,049, Example 1, Step C.

Preparation 3: 3-[1-(4-Chlorobenzy1)-3-(t-buty1thio)-5-hydroxyindo1-2-y1]-2,2-dimethy1-propanoic acid

To a mixture of LiH (12.6 g) and HMPA (105 5 mL) in DMF (1050 mL) at 0°C was added 2-methy1-2propanethiol (178 mL). The mixture was stirred at room temperature for 30 min, then 3-[1-(4-chlorobenzy1)-3-(t-buty1thio)-5-methoxyindo1-2-y1]-2,2dimethylpropanoic acid methyl ester (150 g) (EP 10 419,049, Example 1, Step A) in DMF (450 mL) was added slowly. The mixture was slowly heated to 150°C and kept at that temperature for 18 hours. After cooling to room temperature, the supernatant layer was decanted and the residue dissolved in H20 and 15 acidifed with 1N HCl, extracted twice with Et₂O, washed twice with brine, dried over MgSO4, filtered and evaporated to dryness to provide the title compound.

20 <u>Preparation 4:</u> 3-[1-(4-Chlorobenzy1)-3-(t-buty1thio)-5-hydroxyindo1-2-y1]-2,2-dimethy1-propanoic acid allyl ester

The compound from Preparation 3 (150 g) was dissolved in DMF (1.2 L) then the solution was cooled in an ice-water bath. To this solution was added K₂CO₃ (138 g) portionwise and the mixture was left to stir for 30 min. Then allyl bromide (162 g) was added, the ice bath removed, and the mixture stirred for 18 hours. To the mixture was added aqueous NH₄Cl and it was extracted with Et₂O. The organic layer was washed with H₂O and brine, dried over MgSO₄, filtered, and evaporated to dryness. Purification by silica get chromatography afforded the title compound; m.p. 150-151°C.

WO 94/13662 PCT/CA92/00547

- 51 -

EXAMPLE 1

3-[1-(4-Chlorobenzy1)-3-(t-buty1thio)-5-(pyridin-2-ylmethyoxy)indo1-2-yl]-2.2-dimethylpropanoic acid

5

- 10 Cs₂CO₃ (202 mg), and 2-picolyl chloride (47 mg) were heated at 65°C for 4 hours in a mixture of DMF (3 mL) and CH₃CN (3 mL). After cooling to room temperature, H₂O was added to the mixture which was then extracted twice with EtOAc. The organic extract was then
- washed twice with brine, dried over MgSO4, filtered,
 and evaporated to dryness. The residue was purified
 by silica gel chromatography using 25% EtOAc in
 hexane as eluent to afford the title compound which
 was used as such in the next step.

20

The compound from Step 1 (133 mg) was

- 25 dissolved in THF (5 mL), MeOH (3 mL), and 1N LiOH
 (1.2 mL) and heated at 55°C for 1 hour. After
 cooling to room temperature, the mixture was
 acidified with 1N HC1 and extracted with EtOAc. The
 organic layer was washed with brine, dried over
 - MgSO₄, filtered, and evaporated to dryness. The residue was swished in ether/hexane, affording the title compound as a white solid, m.p. 186.2-187.7°C.

Analysis: Calc'd: C, 67.08; H, 6.19; N, 5.22 Found: C, 66.63; H, 6.16; N, 5.13.

30

EXAMPLE 2

3-[1-(4-Chlorobenzy1)-3-(t-buty1thio)-5-(5-methoxypyridin-2-y1methoxy)indo1-2-y1]-2,2-dimethy1propanoic
acid

Step 1: 3-[1-(4-Chlorobenzy1)-3-(t-butylthio)-5-(5methoxypyridin-2-ylmethoxy)indo1-2-y1]-2,2dimethylpropanoic acid methyl ester

Following the same procedure as in Example

1, Step 1, but substituting 3-[1-(4-chlorobenzy1-3(t-buty1thio)-5-hydroxyindo1-2-y1]-2,2-dimethy1propanoic acid methy1 ester (Preparation 2) for
3-[1-(4-chlorobenzy1-3-(t-buty1thio)-5-hydroxyindo12-y1]-2,2-dimethy1propanoic acid ally1 ester and
5-methoxy-2-picoly1 chloride for 2-picoly1chloride
gave the title compound as a white solid.

Step 2: 3-[1-(4-Chlorobenzy1)-3-(t-buty1thio)-5-(5methoxypyridin-2-ylmethoxy)indol-2-yl]-2,2dimethylpropanoic acid

Following the same procedure as Example 1, Step 2, but using 3-[1-(4-chlorobenzy1)-3-(t-buty1-thio)-5-(5-methoxypyridin-2-ylmethoxy)indo1-2-yl]-

25 2,2-dimethylpropanoic acid methyl ester (Step 1) as starting material afforded the title compound as a white solid.

¹H NMR (250 MHz, CDCl₃): δ 1.20 (s, 9H); 1.23 (s, 6H); 3.30 (s, 2H); 3.88 (s, 3H); 5.19 (s, 2H); 5.35 (s, 2H); 6.7-7.5 (m, 9H); 8.33 (d, 1H).

WO 94/13662 PCT/CA92/00547

- 53 -

EXAMPLE 3

3-[1-(4-Chlorobenzy1)-3-methy1-5-(5-phenylpyridin-2-y1methoxy)indol-2-y1]-2.2-dimethylpropanoic acid

5

A solution of methyl 3-[1-(4-chlorobenzyl)
3-methyl-5-hydroxyindol-2-yl]-2,2-dimethylpropanoate
(180 mg) (Preparation 1), 5-phenyl-2-picolyl bromide
(138 mg), K₂CO₃ (84 mg) and Cs₂CO₃ (30 mg) in DMF (5
mL) wase stirred at r.t. under nitrogen for 48
hours. The mixture was poured onto 1N HCL, extracted

- 2x EtOAc, washed 2x brine, dried (MgSO₄), and
 evaporated to dryness. The residue was purified by
 column chromatography (hexane/EtOAc 5:1 then 1:1) to
 give the title compound.
- 20 <u>Step 2</u>: 3-[1-(4-Chlorobenzyl)-3-methyl-5-(5-phenyl-pyridin-2-ylmethoxy)indol-2-yl]-2,2-dimethyl-propanoic acid

A solution of the ester (173 mg), from Step 1, in 2 mL THF, 2 mL MeOH, and 1 mL 1N LiOH was

25 heated to 80°C under nitrogen for 16 hours. The solution was cooled, poured onto 1N HC1, extracted 2x

EtOAc, washed 2x brine, dried (MgSO₄), and evaporated. The residue was chromatographed on silica gel (eluting with hexane/EtOAc_1:2) to give

the title compound as a solid; m.p. 177-179°C.

WO 94/13662

- 54 -

EXAMPLE 4

3-[1-(4-Chlorobenzy1)-3-methy1-5-(6-phenylpyridin-2-ylmethoxy)indo1-2-yll-2,2-dimethylpropanoic acid

5

Following the procedure described in Example 3, Steps 1-2, but substituting 6-pheny1-2-picoly1 chloride for 5-pheny1-2-picoly1 bromide, the title compound was obtained as a solid; m.p. 137-139°C.

10

EXAMPLE 5

Anal. for $C_{33}H_{38}N_2O_4SC1Na ext{-}2H_2O$

Calc'd C: 60.67; H: 6.48; N: 4.29.

15 Found C: 61.04; H: 6.22; N: 4.32.

EXAMPLE 10

m.p. = 209-211°C.

20

EXAMPLE 11

3-[1-(4-Chlorobenzy1)-3-(t-butylthio)-5-(pyrazin-2-ylmethoxy)indol-2-yll-2.2-dimethylpropanoic acid

25

30

Step 1: 2-Chloromethylpyrazine

To a solution of 2-methylpyrazine (4.7 g) in carbon tetrachloride (200 mL) were added N-chloro-succinimide (8 g) and benzoyl peroxide (0.6 g). The mixture was brought to reflux with two 150-watt spotlights and irradiated for 4.5 hours. The mixture was then cooled to room temperature, evaporated to

WO 94/13662 PCT/CA92/00547

- 55 --

dryness, and chromatographed on flash silica gel using a mixture of ethyl acetate:toluene (1:4) as eluant to give the title compound as an oil, which was stored at 78°C as a solid.

5 l_{H} NMR (Ace- d_{6}): δ 4.8 (2H, s), 8.6 (2H, s), 8.8 (1H, s).

Step 2: Methy1 3-[1-(4-chlorobenzy1)-3-(t-buty1thio)-5-(pyrazin-2-ylmethoxy)indo1-2-y1]2.2-dimethylpropanoate

10

To a solution of 3-[1-(4-chlorobenzy1)-3-(t-butylthio)-5-hydroxyindo1-2-y1]-2,2-dimethyl propanoate (EP 419,049, March 27, 1991, Example 1, Step C) (422 mg) in acetonitrile (5 mL) were added

- solid Cs₂CO₃ (456 mg) and 2-chloromethylpyrazine (117
 mg) from Step 1. The mixture was stirred at room
 temperature for 18 hours. The mixture was poured
 into 25% aqueous ammonium acetate (50 mL), extracted
 with ethyl acetate (2x 50 mL), washed with brine (50
 - mL), dried (MgSO₄), and evaporated to dryness. The residue was chromatographed on flash silica gel using ethyl acetate:toluene (15:85) as eluant to give the title compound as a white solid: m.p. 141-142°C.
- 25 <u>Step 3</u>: 3-[1-(4-Chlorobenzyl)-3-(t-butylthio)-5-(pyrazin-2-ylmethoxy)indo1-2-yl]-2,2dimethylpropanoic acid

The compound from Step 2 (416 mg) was hydrolysed by dissolving it in THF (4 mL), MeOH (2

mL), and 2N LiOH (1.5 mL). The solution was heated at 70°C for 5 hours. The mixture was cooled to room

temperature, diluted with H₂O (50 mL), acidified with glacial acetic acid to pH 5, and diluted with 25% aqueous ammonium acetate (50 mL). The mixture was extracted with ethyl acetate (50 mL), washed with brine (50 mL), dried (MgSO₄), and evaporated to dryness to provide the title compound as a white solid; m.p. 173-175°C.

EXAMPLE 12

10

5

m.p. = 203-204°C.

EXAMPLE 13

m.p. = 238-240°C.

EXAMPLE 14

m.p. = 180-181°C.

20

EXAMPLE 15

 $m.p. = 194-196 \, ^{\circ}C.$

25

EXAMPLE 16

m.p. = 137-139°C.

- 57 -

WHAT IS CLAIMED IS:

1. A compound of the formula I:

5

Het-
$$X^4$$
 R^5
 $CR^{11}R^{11})_n-Y_m-(CR^{11}R^{11})_p-Q$

10

I

wherein:

Het is ArR¹R²;

15

- Ar is a monocyclic aromatic 5- or 6-membered ring containing 1 to 3 N atoms, and the N-oxides thereof;

 R^5 is hydrogen, $-CH_3$, CF_3 , -C(0)H, X^1-R^6 or X^2-R^7 ;

 R^6 and R^9 are independently alkyl, alkenyl, $-(CH_2)_uPh(R^{10})_2$ or $-(CH_2)_uTh(R^{10})_2$;

 R^7 is -CF₃ or R^6 ;

 R^8 is hydrogen or X^3-R^9 ;

- each R¹¹ is independently hydrogen or lower alkyl, or two R¹¹'s on same carbon atom are joined to form a cycloalkyl ring of 3 to 6 carbon atoms;
- 10 R^{12} is hydrogen, lower alkyl or $-CH_2R^{21}$;
 - R^{13} is lower alkyl or $-(CH_2)_rR^{21}$;
 - R^{14} is $-CF_3$ or R^{13} ;

15

- R¹⁵ is hydrogen, -COR¹⁶, R¹³, or two R¹⁵'s on the same nitrogen may be joined to form a monocyclic heterocyclic ring of 4 to 6 atoms containing up to 2 heteroatoms chosen from 0, S, or N;
- R¹⁶ is hydrogen, -CF₃, lower alkyl, lower alkenyl, lower alkynyl or -(CH₂)_rR²¹;
- 25 R^{17} is $-(CH_2)_s-C(R^{18}R^{18})-(CH_2)_s-R^{19}$ or $-CH_2CONR^{15}R^{15}$;
 - R¹⁸ is hydrogen or lower alkyl;

PCT/CA92/00547 WO 94/13662

- 59 -

a monocyclic or bicyclic heterocyclic R^{19} is a) ring containing from 3 to 9 nuclear carbon atoms and 1 or 2 nuclear hetero-atoms selected from N, S or O and with each ring in the heterocyclic 5 radical being formed of 5 or 6 atoms, or the radical W-R²⁰; b) R^{20} is alkyl or -COR²³; 10 phenyl substituted with 1 or 2 R²² groups; R^{21} is R^{22} is hydrogen, halogen, lower alkyl, lower alkoxy, lower alkylthio, lower alkylsulfonyl, lower alkylcarbonyl, -CF3, 15 -CN, $-NO_2$ or $-N_3$; alkyl, cycloalkyl, or monocyclic R23 is monoheterocyclic ring; 20 the residual structure of a standard amino R^{24} is acid, or \mathbb{R}^{18} and \mathbb{R}^{24} attached to the same N can cyclize to form a proline residue; 25 0 or 1; m is 0 to 3; n is 1 to 3 when m is 1; p is 0 to 3 when m is 0; p is 0 to 2; r is 0 to 3; s is 0 to 2; t is .0. to 3;

u is

```
0, S or NR^{15};
      Wis
                   0 or NR<sup>15</sup>;
      X<sup>1</sup> is
                   CO, CR^{11}R^{11}, S, S(0), or S(0)<sub>2</sub>;
      x<sup>2</sup> is
      X^3 is CO, CR^{11}R^{11}, S(0)_2, or a bond;
                   CH=CH, CH<sub>2</sub>-Y<sup>1</sup>, or Y<sup>1</sup>-CH<sub>2</sub>;
      X<sup>4</sup> is
                   x^1 or x^2;
      Y is
      Y<sup>1</sup> is
                   0, S, S(0)_2, or CH_2;
                   -CO_2R^{12}, -CONHS(0)_2R^{14}, -NHS(0)_2R^{14},
      Q is
                   -S(0)_2NHR^{15}, -CONR^{15}R^{15}, -CO_2R^{17},
10
                   -CONR<sup>18</sup>R<sup>24</sup>, -CR<sup>11</sup>R<sup>11</sup>OH, or 1H- or
                   2H-tetrazo1-5-y1;
      or a pharmaceutically acceptable salt thereof.
15
                   2. A compound of Claim 1 wherein X^4 is
      CH_2-Y^1 and Y^1 is 0;
      or a pharmaceutically acceptable salt thereof.
20
                   3. A compound of Claim 1
       wherein:
      \mathbb{R}^1, \mathbb{R}^2, \mathbb{R}^3, and \mathbb{R}^4 are hydrogen;
      R^5 is X^2-R^7;
25
      R^7 is R^6;
      R^8 is R^9;
      R<sup>10</sup> is hydrogen or halogen;
      m is 0;
30
      n is 1 to 3;
```

u is 0 in R⁶ and 1 in R⁹;

- 61 -

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x^2 is CR^{11}R^{11} or S;
     X^4 is CH_2-Y^1;
      Y<sup>1</sup> is 0; and
      Q is -CO_2R^{12};
 5
      or a pharmaceutically acceptable salt thereof.
               4. The compound of Claim 1
      wherein:
 10
     R^1, R^2, R^3, and R^4 are hydrogen;
     R^5 is X^2-R^7;
     R^7 is R^6;
     R^8 is R^9:
     R<sup>10</sup> is hydrogen or halogen;
 15
m is 0;
     n is 1 to 3;
     u is 0 in R<sup>6</sup> and 1 in R<sup>9</sup>;
     X^2 is CR^{11}R^{11} or S;
     X^4 is CH_2-Y^1;
 20
     Y^1 is 0; and
      Q is 1-H- or 2H-tetrazo1-5-y1;
 or a pharmaceutically acceptable salt thereof.
```

5. A compound of Claim 1 of the formula Ia:

$$R^{1}R^{2}-Ar-X^{4}$$
 $CH_{2}-Y-(CR^{11}R^{11})_{p}-Q$ R^{8} Ia

wherein the substituents are as follows:

	Ex	R ¹ /R ²	Ar	χ ⁴ .	R ⁵	R ^B	Y-(CR ¹¹ R ¹¹) _p -Q
•	No.						
15	1	н/н	pyrid-2-yl	СН ₂ 0	S-t-Bu	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H
	2	5-OMe/H	pyrid-2-yl	СН ₂ 0	S—t—Bu	CH ₂ Ph-4-C1	C(Me)2CO2H
	3	5-Ph/H	pyrid-2-yl	СН ₂ 0	Me	CH ₂ Ph-4-C1	C(Me)2C02H
	4	6-Ph/H	pyrid-2-yl	сн ₂ 0	Me	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H
	5	3,5-Me ₂ /4-OMe	pyrid-2-yl	CH ₂ 0	S-t-Bu	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H
20	6	н/н	pyrazin-2-yl	CH ₂ S	Me	CH ₂ Ph-3-F	C(Me)2CONH
							\$(0) ₂ Me
	7	5-C1/H	pyrimidin-2-yl	CH2CH2	COCH ₂ -t-Bu	CH ₂ Ph-4-CN	C(Me) ₂ CHOHMe
	8	1-Me/H	imidazol-2-yl	CH ₂ 0: =	CH ₂ -t-Bu	CH ₂ P.h-4=C1	C(Me)2C02H
	9	1-Me/H	imidazol-4-yl	сн ₂ 0	CH ₂ -t-Bu	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H
25	10	6-C1/H	pyrid-2-yl	сн ₂ 0	St-Bu	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H
	11	H/H	pyrazin-2-yl	CH ₂ 0	S-t-Bu	CH2Ph-4-C1	C(Me)2C02H
	12	H/H	pyrid-3-yl	CH ₂ O	S-t-Bu	CH2Ph-4-C1	C(Me)2C02H
	13	H/H	pyrid-4-yl	CH ₂ 0	S-t-Bu	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H
	14	H/H	pyrazin-2-yl	сн ₂ 0	S-t-Bu	CH ₂ Ph-4-S-t-Bu	C(Me) ₂ C0 ₂ H
30	15	H/H	pyrid-2-yl	сн ₂ 0	COCH ₂ -t-Bu	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H
	16	H/H	pyrid-2-yl	сн ₂ 0	H	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H
	17	H/H	pyrid-2-yl	сн ₂ 0	CH ₂ —t-Bu	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H
	18	H/H	pyrid-2-yl	СH ₂ 0	СН ₂ -с-Ви	CH ₂ Ph-4-C1	C(Me) ₂ C0 ₂ H

PCT/CA92/00547 WO 94/13662

- 63 -

6. A pharmaceutical composition comprising a therapeutically effective amount of a compound of Claim 1 and a pharmaceutically acceptable carrier.

A pharmaceutical composition of Claim 6 5 additionally comprising an effective amount of a second active ingredient selected from the group consisting of non-steroidal anti-inflammatory drugs; peripheral analgesic agents; cyclooxygenase inhibitors; leukotriene antagonists; leukotriene 10 biosynthesis inhibitors; H2-receptor antagonists; antihistaminic agents; prostaglandin antagonists; thromboxane antagonists; thromboxane synthetase inhibitors; and ACE antagonists.

15

8.—A method of preventing the synthesis, the action, or the release of SRS-A or leukotrienes in a mammal which comprises administering to said mammal an effective amount of a compound of Claim 1.

20

9. A method of treating asthma in a mammal comprising administering to a mammal in need of such treatment a therapeutically effective amount of a compound of Claim 1.

10. A method of treating inflammatory diseases of the eye in a mammal which comprises administering to a mammal in need of such treatment a therapeutically effective amount of a compound of

Claim 1.

INTERNATIONAL SEARCH REPORT International Application No

PCT/CA 92/00547

I. CLASSI	FICATION OF SUBJE	CCT MATTER (if several classification sy	mbols apply, indicate all) ⁶	
	to International Patent . 5 CO7D401/	Classification (IPC) or to both National Cl 12; A61K31/405;	assification and IPC CO7D403/12	
II. FIELDS	SEARCHED			
		Minimum Docume	ntation Searched ⁷	
Classificat	tion System		Classification Symbols	
Int.C1	. 5	CO7D ; A61K		
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		D TO BE RELEVANT ⁹ ocument, ¹¹ with indication, where appropria	12	Dalaman Claim No 13
Category °	Citation of De	ocument, " with indication, where appropria	ite, of the relevant passages 2	Relevant to Claim No.13
X	CORPORA 21 May cited in			1-7
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"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention and the principle or theory underlying the invention and the principle or theory underlying the invention are document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed TV. CERTIFICATION "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention cannot be considered novel or cannot be considered to involve an inventive step when the document is combined with one or more other such document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. "A" document member of the same patent family				
Date of the Actual Completion of the International Search O7 APRIL 1993 28. 04. 69				
International Searching Authority Signature of Authorized Officer				
EUROPEAN PATENT OFFICE GETTINS M.P.				

International Application No

III. DOCUME	NTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)	Relevant to Claim No.
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Y	EP,A,O 419 049 (MERCK FROSST CANADA INC.) 27 March 1991 cited in the application see claim 1	1-7
Y	EP,A,O 279 263 (ABBOTT LABORATORIES) 24 August 1988 cited in the application see claim 1	1-7
Y	EP,A,O 275 667 (MERCK FROSST CANADA INC.) 27 July 1988 cited in the application see claim 1	1-7
Y	EP,A,O 166 591 (MERCK FROSST CANADA INC.) 2 January 1986 cited in the application see claim 1	1-7
Y	WO,A,9 203 132 (ABBOTT LABORATORIES) 5 March 1992 see page 3, line 9; claim 1	1-7
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ANNEX TO THE INTERNATIONAL SEARCH REPORT ON INTERNATIONAL PATENT APPLICATION NO.

CA 9200547 SA 68116

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on

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ANNEX TO THE INTERNATIONAL SEARCH REPORT ON INTERNATIONAL PATENT APPLICATION NO.

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